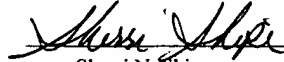


CERTIFICATE OF MAILING BY "FIRST CLASS MAIL"

I hereby certify that this correspondence is being deposited with the United States Postal Service as first class mail in an envelope addressed to:
Assistant Commissioner for Patents, Washington, D.C. 20231, on July 27, 1999.


Sherri N. Shipe



IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In the application of:

S.N. VAUGHN

Serial No.: 08/943,399

Filing Date: October 3, 1997

For: METHOD FOR INCREASING LIGHT
OLEFIN YIELD BY CONVERSION OF
A HEAVY HYDROCARBON
FRACTION OF A PRODUCT TO
LIGHT OLEFINS

Examiner: In Suk Bullock

Group Art Unit: 1764

DECLARATION UNDER 37 C.F.R. 1.131

Assistant Commissioner for Patents
Washington, D.C. 20231

Dear Sir:

1. This declaration is to establish completion of the invention in this application in the United States, at a date prior to July 22, 1997, the effective date of U.S. Patent No. 5,914,433.

2. The person making this declaration is the inventor.

3. To establish the date of completion of the invention of this application, attached is a copy of Patent Memorandum PM97019 redacted form. PM97019 includes a complete description of the claimed invention. In addition, PM97019 includes actual data showing an actual completion or reduction to practice of the invention.

4. I have reviewed what I believe to be an exact photocopy of the original Patent Memorandum PM97019, which bears the signature of D.Y. Ou as having witnessed, read and understood the document on a date prior to July 22, 1997. D.Y. Ou is presently Engineering Associate at Exxon Chemical Company, and was employed by Exxon Chemical Company at the time of having witnessed, read and understood Patent Memorandum PM97019.

5. I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Dated: 7/22/99

By:

Stephen N. Vaughn
Stephen N. Vaughn

PATENT MEMORANDUM

INVENTOR(S)

Stephen N. Vaughn

DIVISION MEMO NO

CLASS

EXXON CHEMICAL COMPANY JUL 28 1999

PATENT COMMITTEE

STATUS OF INVENTION (EXISTING OR CONTEMPLATED EXPERIMENTAL, COMMERCIAL OR PLANT SCALE USE)

Experimental

DATE OF FIRST USE, PUBLICATION, AND/OR SALE

EXACT DESCRIPTIVE TITLE

Process to convert C₄ + Stream in Methanol to Olefins Process

BRIEF ABSTRACT OF INVENTION

Hydrocarbons heavier than propane are separated from a Methanol to Olefins product stream and converted to generate additional ethylene and propylene either in a separate reactor or in the MTO reactor.

FOR LAW-TECHNOLOGY DEPARTMENT USE ONLY

DATE RECEIVED

ATTORNEY'S INIT.

☐ YES☒ NO

CONTRACT NO.(S)

FIRST WRITTEN DESCRIPTION

HEREWITH OR FILED AT

DATE

Notebook No 19763 p 59

DATE OF FIRST TEST

WHERE RECORDED? (E.G. NOTEBOOK & PAGE)

Notebook 19763 p 59

FIRST SKETCH OR DRAWING

HEREWITH OR FILED AT

DATE

Herewith

COMPANY REPORTS, FILE MEMOS, PMS RELATED TO INVENTION (IDENTIFY)

none

FIRST DISCLOSURES TO OTHERS

TO WHOM AND WHERE

DATE

TJ Chen BCIT, Baytown Tx

IS MODEL OR PROTOTYPE AVAILABLE?

☐ YES☒ NO

DETAILED DESCRIPTION OF INVENTION (SEE INSTRUCTIONS ON REVERSE SIDE)

The Methanol-to-Olefins (MTO) process as currently envisioned is remarkably selective to ethylene and propylene. Depending upon the process conditions used, C₂ = + C₃ = yields of 80-90% are expected to be achieved. Even with this highly selective process however, approximately 5-10% of the organic portion of the methanol feed ends up as relatively low value C₄ + olefins.

To make better use of this stream, it has been discovered that the C₄ + olefins can be recycled to produce additional ethylene and propylene. As an example of this, Table 1 illustrates that when 1-butene is reacted over an MTO catalyst, propylene, ethylene and C₅ + are generated as products. It is believed that a significant fraction of the C₄ + stream will ultimately form coke. Tests using 1-butene as a surrogate for the C₄ + stream show a selectivity of approximately 10.1 % to coke. If the other heavy products react in the same way, the typical yield of approximately 12% C₄ + products in unrecycled MTO can be replaced with 1.4% C₂ = and 8.0% C₃ = and 1.5 % coke when the C₄ + stream is recycled to extinction in the MTO reactor. Other yields will be realized if the C₄ + stream is converted in a separate reactor as described below. The extra coke can be burned in the catalyst regenerator for heating value and the plus C₂ = and C₃ = provides additional high value products.

The recycle can be accomplished in at least two ways. As Figure 1 shows schematically, the C₄ + products can be separated from the more valuable C₂ = and C₃ = cryogenically and the entire C₄ + stream (or part of that stream) can be combined with the methanol feedstream. [Note: The cryogenic separation shown in Figs 1 and 2 is intended to be generic and does not preclude the inclusion of additional separation steps necessary to produce ethylene and propylene of the required purity, while producing a stream composed substantially of C₄ + products that can be further converted as described in this invention]. After reaction in the MTO reactor, the heavy stream comprising C₄ + products produced from both methanol and previously recycled C₄ + products can be again separated and recycled; this process can be continued until no net C₄ + products are produced or alternatively, sized to produce a specific amount of C₄ + products.

(Continue Description on Additional Sheet(s))

INVENTOR(S)	
SIGNATURE (FIRST NAME IN FULL)	SIGNATURE (FIRST NAME IN FULL)
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DATE	DATE
INVENTOR EMPLOYMENT CONTRACT	INVENTOR EMPLOYMENT CONTRACT
CITIZENSHIP:	CITIZENSHIP:
<input type="checkbox"/> YES <input checked="" type="checkbox"/> NO	<input type="checkbox"/> YES <input checked="" type="checkbox"/> NO
<input type="checkbox"/> U.S.A. <input type="checkbox"/> OTHER	<input type="checkbox"/> U.S.A. <input type="checkbox"/> OTHER
EXXON CHEMICAL CO. DIVISION:	EXXON CHEMICAL CO. DIVISION:
OTHER:	OTHER:

11/11/10

PATENT MEMORANDUM

Page No. 2

INVENTOR(S)

Stephen N. Vaughn

EXACT DESCRIPTIVE TITLE

Process to convert C₄ + Stream in Methanol to Olefins Process

(For Continued Detailed Description of Invention)

In this mode of operation, one additional benefit to the overall MTO process is the inclusion in the MTO reactor of a certain amount of endothermic C₄⁺ cracking reactions which will help to counteract the exothermic MTO reactions.

Alternatively, as illustrated schematically in Fig 2, after cryogenic separation the C₄⁺ stream can be fed into a relatively small auxiliary reactor. By using a separate reactor, process conditions and catalysts can be selected to optimize the production of the desired products. As examples of the benefits of reacting the C₄⁺ stream under conditions different from optimal MTO conditions, Table 2 illustrates the effect of using ZSM-5. ZSM-5 is not generally considered to be a good MTO catalyst but it has a higher activity for butene conversion than does the SAPO-34 catalyst.

The current invention encompasses the idea of separating C₄⁺ products from an MTO product stream (by means well known to those skilled in the art) and directing this separated stream to either 1) the MTO conversion reactor or 2) a separate C₄⁺ conversion reactor. In the first case the conversion of C₄⁺ will be achieved on the MTO conversion catalyst at conditions favorable for MTO. In the second case, a number of catalysts can be employed including SAPO-34, -44, -18, and -17, ZSM-5 or medium pore zeolites. The process conditions in the second case can be optimized independently of the MTO conversion reactor and are expected to be in the ranges of

WHSV 0.1 to 100 hr⁻¹
Temperature 300-650 °C
Pressure 15-100 psia
Dilution none or steam

Again depending upon the catalyst employed, on-line or periodic regeneration of the C₄⁺ conversion catalyst may be provided by means well known to those skilled in the art.

In either case it may be advantageous to provide for some C₄⁺ product purge --to provide a ready source of fuel gas for plant use, C₄⁺ product for sale or to limit the size of the recycle stream.

25 psig /WHSV 120 hr⁻¹/ pure 1-butene/450 °C

Table 1

Selectivity (wt%, excluding coke and water)

Feed	Conversion	CH ₄	C ₂ =	C ₃ =	C ₄	C ₅ +
1-C ₄ =	34.23	0.10	10.57	66.29	N/A	21.74
MeOH	100.00	0.86	51.06	34.19	9.29	2.08

25 psig /WHSV 120 hr⁻¹/ pure 1-butene/450 °C

Table 2

Selectivity (wt%, excluding coke and water)

Catalyst	Conversion	CH ₄	C ₂ =	C ₃ =	C ₄	C ₅ +
SAPO-34	34.23	0.10	10.57	66.29	N/A	21.74
ZSM-5	73.10	0.21	22.31	53.03	N/A	19.91

INVENTOR(S)

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PATENT MEMORANDUM

Page No. 3

(For Continued Detailed Description of Invention)

INVENTOR(S)

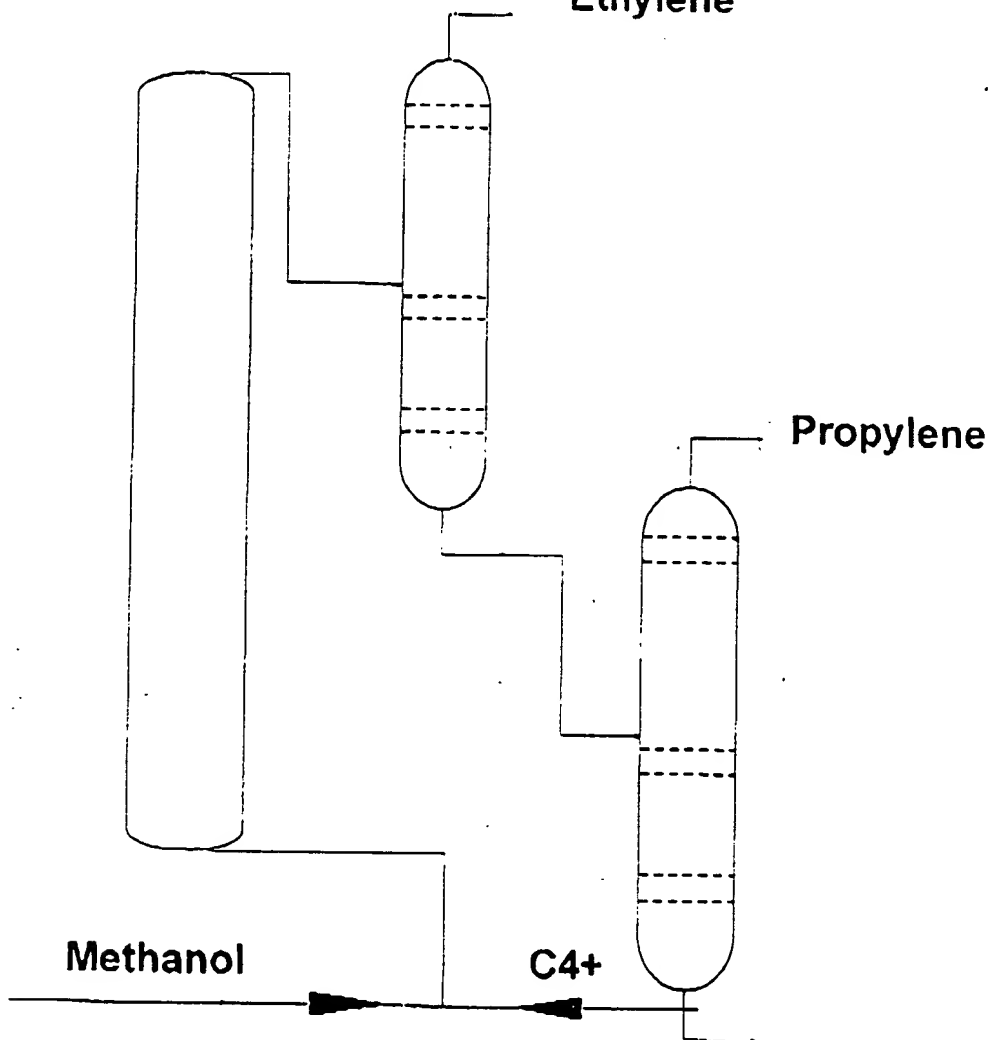
S. N. Vaughn

EXACT DESCRIPTIVE TITLE

Process to convert C4 + Stream in Methanol to Olefins Process

Ethylene

Fig 1



MTO Reactor

Cryogenic Separation

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(For Continued Detailed Description of Invention)

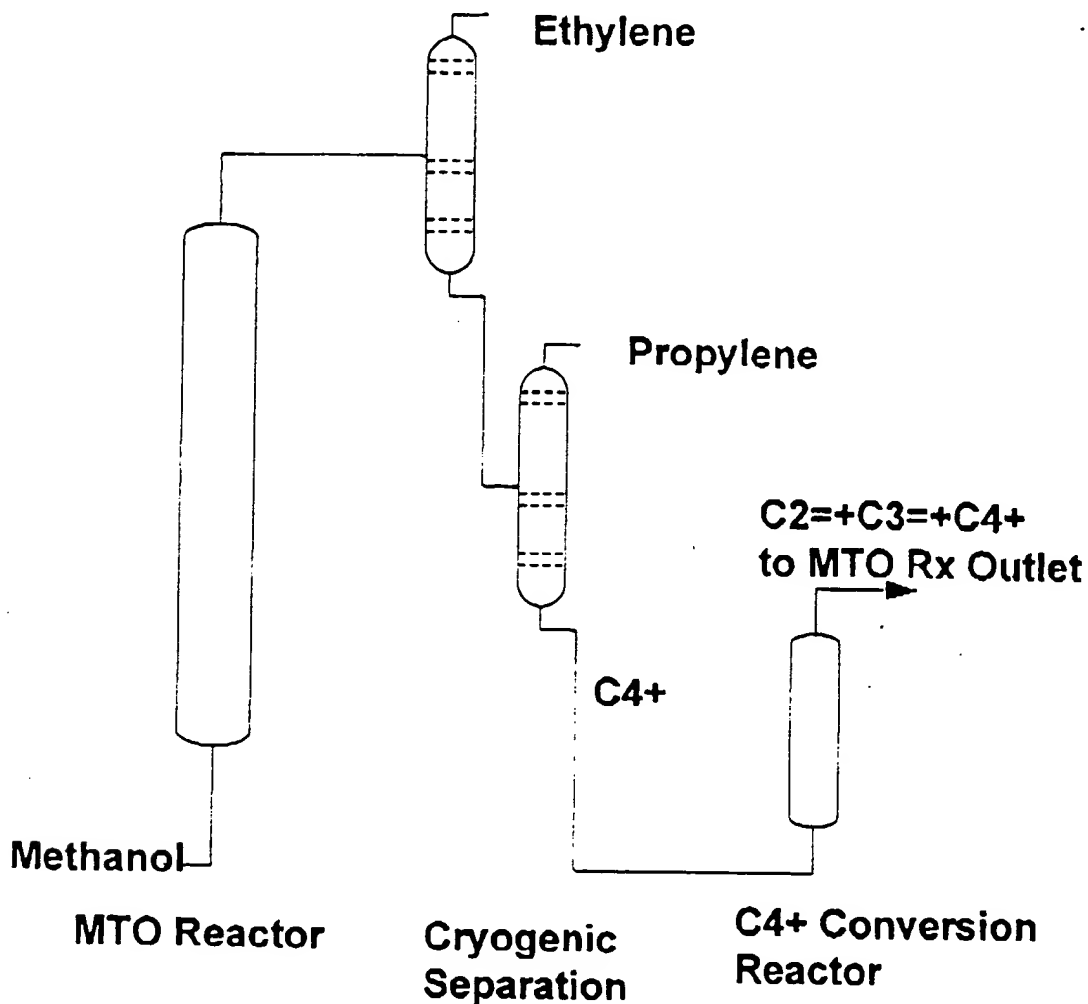
INVENTOR(S)

S. N. Vaughn

EXACT DESCRIPTIVE TITLE

Process to convert C4 + Stream in Methanol to Olefins Process

Fig 2



INVENTOR(S)

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